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Characterization of petroleum-based plastics and their absorbed trace metals from the sediments of the Marina Beach in Chennai, India

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Abstract

Background: Pollution by marine plastic debris becomes an increasing global concern. Characteristics of petroleum-based plastics pollution and its absorbed metals in Marina Beach, Chennai, in India are not reported yet. In this study, we first characterized the petroleum-based plastic debris and quantified the concentration of arsenic (As), chromium (Cr), copper (Cu), and lead (Pb) in plastics from the Marina Beach.

Results: A total of 2275 samples of the sediments were collected from three different locations across the Marina Beach. Our results showed that plastics debris mainly were three shapes as fiber, sphere, and fragment. The plastic fragments were mechanically eroded and chemically weathered observed by the scanning electron microscope (SEM). Five types of polymers were identified, including low-density polyethylene (LDPE), polyethylene terephthalate (PET), polypropylene (PP), polyamide (PA), and polycarbonate (PC). Their absorbed trace metals, such as Cr, As, Pb, and Cu, were determined by the inductively coupled plasma mass spectrometry (ICP-MS). The level of As was high in most plastics. However, the highest levels of Pb, Cu, and Cr were in the PA, PP, and PA, respectively, implying the adsorption preference of plastics from different sources.

Conclusion: Taken together, this study first characterized petroleum-based plastics and their absorbed trace metals in the sediments of the Marina Beach, indicating that petroleum-based plastics acted as the vectors for the transportation of metal elements in the marine biome.

Keywords: Plastic, Marina Beach, Chennai, Trace metal, Vector

Background

Petroleum-based plastics are artificial organic polymers, obtained from natural gas or oil, and utilized in contemporary society in every aspect of daily life [1-3]. Global petroleum-based plastic production in

2015 reached 322 million tons, while in 1950, the production was 1.7 million tons [4]. During the twentieth century urbanization in the vicinity of coastal environments, the utilization of petroleum-based plastic materials in the various applications, especially in the medicinal, industrial area, commercial, and municipal field, has steadily increased and unknowingly or knowingly released into the environment, which leads to a significant environmental problem [5–8]. Petroleum-based plastics may be fragmented, but cannot naturally break down due to their profound stability [9]. Even the

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degradable plastic materials can last for many years, depending on environmental circumstances, such as temperature, oxygen, and the exposure levels of ultraviolet (UV) light [10]. Also, stabilizers and plasticizers are included to enhance the characteristics features and efficiency of petroleum-based plastic materials [11], and during the degradation process, they may be leaked into the environmental system [12].

Generally, the term plastics are referred to larger sized plastic materials ranging from 2.5 cm to 1 m of known items, such as bottles, large containers, plastic toys, and buoys. The term mesoplastics are generally referred to as the size of particles from 5 mm to 2.5 cm [13]. Various researchers have faced practical categories in dividing small plastics (0.1-1 mm) and abundant plastics (1-5 mm) [14]. The primary microplastics are produced in microsize, and the secondary microplastics are generated by disintegrating larger plastic materials to smaller plastic fragments [3]. The accumulation of large plastic debris in the marine ecosystem is because of the severe discharge of plastic-based materials [15], albeit some of the waste plastics are appropriately done by burning or recycling. In 2010, around 4.8 to 12.7 million metric tons of plastics were released into the marine biome [16].

After the plastic debris is thrown out into beaches, it will absorb metals from the surrounding environment [17–22]. The toxic effect may happen when the organism is exposed [23]. Rare studies of a petroleum-based plastic accumulation from the beach sediment in India were carried out [24–26]. Several studies have already shown that organic pollutants are absorbed in plastic debris [14, 27]. Recently, the existence of metals on plastic waste was reported [17, 18, 28]. Thus, the plastic fragments act as stressing agents to the organisms, causing adverse effects after consumption [20].

India is among the world's largest consumers of petroleum-based plastics, which can produce about 5.6 million tons of plastic material annually [29]. Chennai is the capital of Tamil Nadu, located in the south-eastern coastal line of India, 56 km along the coast, which is the fourth most crowded metropolitan city in India [21]. Each year, monsoon carries plastic fragments from the waste piles into the lakes, streams, and rivers, eventually to the beach. Because of human activities, petroleum-based plastics are accumulating on the coastal line, reducing the esthetic attraction, and damaging natural life [1]. The main feasible steps to minimize the environmental impact of petroleum-based plastics are mainly through recovery, treatment, recycling [2]. The qualitative and quantitative study of plastic debris is used to define basic needs for the management of the most plastic waste. By adequately managing anthropogenic waste, economic and ecological effects on marine biomes could be eased [30]. This study first characterized petroleum-based plastics through the coastline of the Marina Beach according to the size, shape, and color, and quantified their absorbed trace metals including As, Cr, Cu, and Pb.

Materials and methods

Study area and sampling

Marina beach is 6 km long and 300 m wide, which is one of the busy beaches that draws about 30,000 tourists a day on weekdays and 50,000 on weekends and holidays [31]. This study was conducted at three sites: (1) S1: Lighthouse (13.039687° N, 80.279378° E); (2) S2: Thiruvalluvar statue (13.054669° N, 80.282343° E); (3) S3: MGR memorial (13.064330° N, 80.285322° E) in the Marina Beach littoral zone (Additional file 1: Fig. S1). From December 1 to December 4, 2017, all samples were collected. Three different locations were chosen at intervals of about 2 km. Measurements were carried out for each location using quadrates 5 m in width and 100 m in length between the high tide and the vegetation line according to the previous study [32]. At each site, three replicates samples were collected. The debris and 5 cm of sand from each site in the quadrats were sieved as described [24, 33, 34]. Plastic debris and natural particles on sieves were placed in zipper bags and transported to the laboratory for the subsequent study.

The optical microscopy analysis

All the samples were observed under microscopic observations carried out using Leica Model SZH10, Japan, 4. Plastic debris was classified by shape and color. The same color and shape plastics were known as the same type [35]. The size of the plastic debris was then measured. The measurements of each irregular fragment by the longest length measured using image processing software. After sorting the samples, plastics were thoroughly washed in deionized water using the ultrasonic bath to eliminate tarry residues and sand, dried at 60 °C for 12 h. The samples were stored at ambient room temperature conditions for further analysis.

The SEM analysis

Five samples of each form of plastics were analyzed using the SEM to understand the surface characteristics of the plastic debris. Samples were fixed on the aluminum SEM stubs on the dual-sided carbon adhesive tabs according to the previous description [36], and were further sputtered with a thin film using the MCM-100P gold coating unit. The film's coating thickness was, on average, 10 nm at 8 to 10 Pa vacuum speeds. Samples were studied at an accelerating voltage of 10 kV using SEM (Carl Zeiss).

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Fourier transform infrared spectroscopy–attenuated total reflectance (ATR–FTIR) and thermogravimetry–differential scanning calorimetry (TGA–DSC)

The polymer composition of 20 components with each shape was identified using the ATR–FTIR technique and TGA–DSC method. The most accurate technique for the identification of plastic chemical composition was infrared spectroscopy [37, 38]. ATR–FTIR was used to analyze plastics. The plastic debris was analyzed using the FTIR spectrometer with a DTGS detector on a Jasco 4700 spectrum. The wavelength ranging from 4000 to 600 cm⁻¹, and 16 scans per sample were conducted at a resolution of 4.0 cm⁻¹. The OPUS 7.5 software was used to compare the polymer spectrum from the database of the most typical polymer and natural material.

A STA6000 (Perkin Elmer) was used to measure TGA–DSC with a flow rate of 20 mL min $^{-1}$. The heating rate at 10 °C min $^{-1}$ with different temperatures was set for dynamic step interruption. The sample was heated in the inert nitrogen atmosphere ranging from 20 to 800 °C with 5 K min $^{-1}$ to suppress oxidation. At 20 °C gradient, the measurement was started.

Analysis of metal concentration in the plastic debris

About 1 g plastic items from each polymer compositions were selected and put into the Teflon microwave digestion tube, and 5 mL of concentrated supra-pure nitric acid was added [39, 40]. The vessels of Teflon were assembled, and the acid digestion step was carried out for 50 min. Teflon vessels were placed into Mars 6 microwave digester. The temperature template was set around for 20 min at 200 °C, the hold

temperature at 200 °C for 15 min, and 15 min for cooling. Once the digestion was completed, the digested sample was transferred into a 50-mL polyethylene centrifugal vial.

Analyses of As, Cr, Cu, and Pb were performed using the ICP-MS (7700 x, Agilent, USA); to maintain inert condition helium gas was induced inside the collision reaction cell. Radiofrequency mainly focused on the ions and later reacted and collided with collision reaction gas. The helium mode was used to filter all spectral interferences including unidentified and discrete sample matrices. The ICP-MS was calibrated using internal standardization achieved by the addition of ⁴⁵scandium, ⁷²germanium, ¹¹⁵indium, and ²⁰⁹bismuth. The operation conditions for the instrument RF power 1400 is with nebulizer gas 0.30 L min⁻¹, inert gas argon 15 L min⁻¹, uptake of samples at the rate of 0.3 mL min ⁻¹. Samples were analyzed through 100 scans per repeat, and 3 repeats per sample were conducted [41]. The detection limit of the instrumental was 1.7 ppt for Cu, 4.3 ppt for Cr, 1.3 ppt for Pb, and 11.9 ppt for As, respectively.

Statistical analysis

The data were analyzed using the XLSTAT software (Addinsoft, USA, 2019) and analyzed on the confidence level with Tukey (HSD) test. The values were indicated as the mean±standard error (SEM) of the replicates in each group. The multivariate analysis, including principal component analysis (PCA), was performed for data analysis.

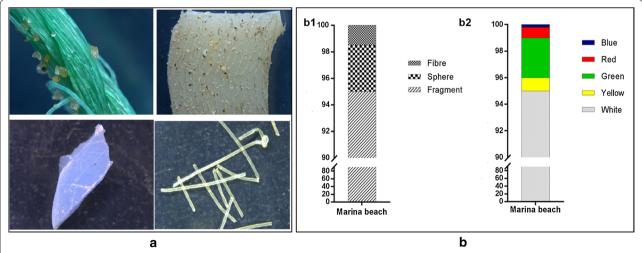


Fig. 1 The shape and color distribution of petroleum-based plastics in this study. **a** Representative images of unwashed petroleum-based plastic debris from the sediment samples from the Marina Beach, Chennai, India. **b** The shape and color distributions of petroleum-based plastic debris on the Marina Beach

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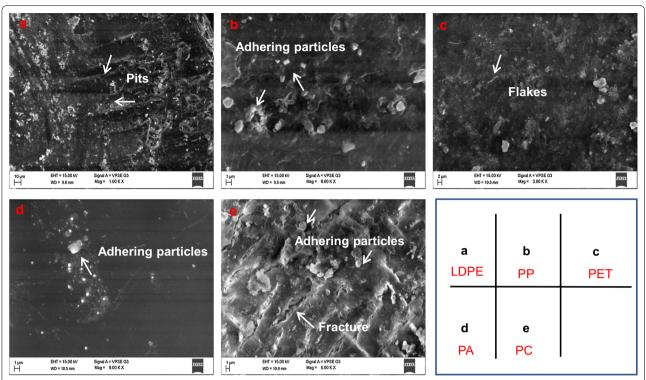


Fig. 2 The texture of chosen petroleum-based plastics. a Pits on the LDPE fragment; b adhering particles on PP fragment; c flakes on PET; d early degradation site and adhering particles on a PA; e adhering particles on PC

Results

Shape, color, and size of petroleum-based plastic

Plastic debris with three shapes were identified as fiber, sphere, and fragment (Fig. 1). There were significantly different plastic debris numbers between sampling stations. The Lighthouse and Thiruvalluvar Statue had the highest (830 items) and lowest (636 items) total numbers of plastics debris (Additional file 1: Fig. S2A and Tables S1, S2). Overall, the size of the plastic ranged from 5.5 to 25 mm (Additional file 1: Fig. S2B). The most found was fragment with about 95%, followed by sphere (3.5) and fiber (1.5), respectively (Fig. 1b1). As shown in (Fig. 1b2), 90% of the total plastic was white; the other color is yellow, green, red, and blue. There was significant weathering of the plastic debris.

Surface texture and aging status of petroleum-based plastics

The result of the SEM images showed that the mechanical erosion and chemical weathering were found in the samples collected in the surface sediments of Marina Beach. Pits, adhering, and flakes were the usual pattern of degradation of plastics (Fig. 2a–e), which was confined to mechanical abrasions through adhering particles and grinding of sand. Besides mechanically created pits and fractures, additives used in plastics, such as colorants,

fillers, stabilizing materials, and softeners, should have provided an intended area for degradation to begin (Fig. 2a–e). Particles with flakes were available locally for foreign objects to become accommodated, thereby making it possible for the polymer to mechanical degradation (Fig. 2b, d, e).

Composition characterization using the ATR-FTIR and TGA-DSC

As shown in Fig. 3, plastic particles of LDPE, PP, PET, PA, and PC were detected using ATR–FTIR, and TGA–DSC. For the LDPE analysis, the strong asymmetric CH₂ stretch of 2913–2911 cm⁻¹, the symmetric CH₂ stretch of 2847 cm⁻¹, and the symmetric deformation of CH₃ were the characteristic spectra of LDPE (Fig. 3a, left panel). There were three bands assigned to CH₂ and CH₃ groups: a band I at 1370 cm⁻¹. The thermal stability of plastic debris showed the degradation in a single-step process, and the abundant temperature for decomposition was about 478 °C for LDPE. DSC results showed that the melting point (Tm) of LDPE is 122 °C (Fig. 3a, right panel).

For the PP analysis, the absorbance spectra were shown at 2950 cm⁻¹, 1455 cm⁻¹, and 1371 cm⁻¹. In the spectrum, the moderate absorption peaks occurred in the

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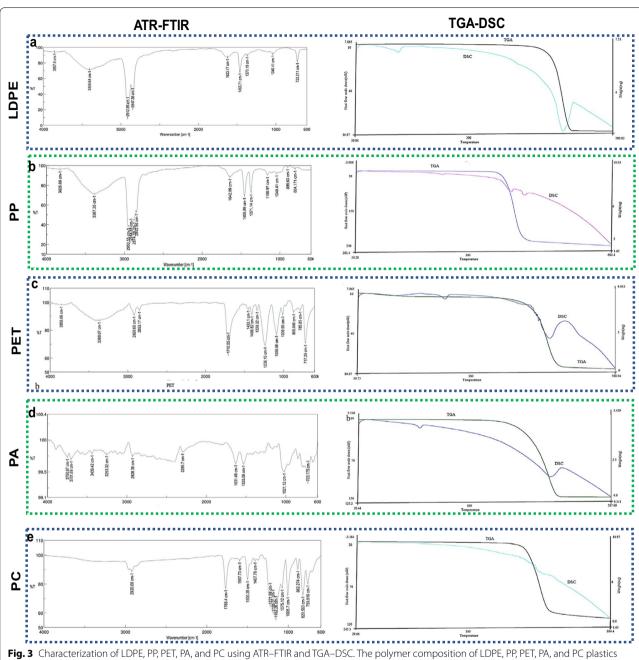


Fig. 3 Characterization of LDPE, PP, PET, PA, and PC using ATR–FTIR and TGA–DSC. The polymer composition of LDPE, PP, PET, PA, and PC plastics debris was characterized using ATR–FTIR, and the peaks of LDPE, PP, PET, PA, and PC were determined by the TGA and DSC under nitrogen atmosphere

spectral band of 1445 cm^{-1} and 1371 cm^{-1} . The absorption characteristics of 804 and 1001 cm⁻¹ were the typical vibrations of terminal unsaturated CH_2 groups in isotactic PP (Fig. 3b).

For the PET analysis, the spectra of PET spectra showed a substantial intensity at 1710 cm⁻¹, which was assigned as (C=O) stretching carbonyl bonds. The peaks at 1406, 866, and 717 cm⁻¹ were the bonds of (C=C),

(=C-H), and (C-H) detected by stretching vibrations. The vibrational bands 1339 cm⁻¹, 1236, 1089, and 1009 cm⁻¹ were assigned to C-H and C-O stretching vibration (Fig. 3c). The PET is thermally stable and can be decomposed at a maximum temperature of about 445.18° C in a single cycle. The DSC results in Fig. 3c suggested that the melting point (Tm) of PET was 219.64 °C.

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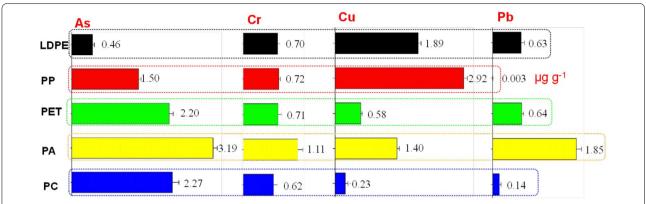


Fig. 4 The concentration of trace metals from different petroleum-based plastic debris in the Marina Beach. The error bars indicated the standard deviation. The unit was $\mu g g^{-1}$

For the PA analysis, $\rm CH_2$ was symmetrically extended, but detected an asymmetric site at 2928 cm⁻¹ with $\rm CH_2$ group vibrations in the polymer chain. The C=O stretching was related to the spectrum at 1631.48 cm⁻¹, but there was a spectrum at 1535 cm⁻¹ due to the N–H band combination (Fig. 3d, left panel). The thermal stability of PA plastic debris showed the degradation in a single-step process, and the maximum temperature for decomposition was 455.18 °C (Fig. 3d, left panel). DSC results showed that the melting point (Tm) of PA was 165.64 °C (Fig. 3d, right panel).

For PC analysis, the ATR–FTIR spectrum showed the vibration at 2920 cm $^{-1}$ and 1768.4 cm $^{-1}$ indicated an aromatic (C–H) and a strong intensity of carbonyl bonds C=O (Fig. 3e, left panel). The thermal stability of plastic debris PC showed the degradation in a single-step process, and the intended temperature for decomposition was found to be 560 °C (Fig. 3e, right panel). The results of DSC showed that the melting point (Tm) of polycarbonate was 391 °C (Fig. 3e, right panel).

The absorption pattern of trace metals in the plastic debris

In this study, the trace metal concentration of plastic debris collected from the Marina Beach is shown in Fig. 4 and Additional file 1: Table S1. The concentrations of As ranged from 3.19 ± 0.09 to $0.46\pm0.06~\mu g~g^{-1}$, with the maximum in the PA and the minimum in the LDP. For the ANOVA and Tukey test, the level of As was different in plastic debris ($p\!=\!0.05$) (Fig. 4, Additional file 1: Fig. S3A and Table S3). The Cr level ranged from 1.11 ± 0.155 to $0.62\pm0.106~\mu g~g^{-1}$ with the maximum in the PA and the minimum in the PC. However, the levels of Cr were identical between polymers except for PA, according to ANOVA and Tukey's test ($p\!=\!0.05$) (Additional file 1: Fig. S3B and Table S4). The level of Cu in the polymers ranged from 2.92 ± 0.190 to $0.23\pm0.084~\mu g~g^{-1}$ with the

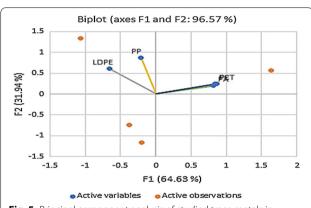


Fig. 5 Principal component analysis of studied trace metals in the petroleum-based plastic debris collected from Marina Beach, Chennai, India

maximum content in the PP and the minimum content in the PC. The difference of Cu levels was significant (Additional file 1: Additional file 1: Fig. S3C and Table S5). The levels of Pb among polymers ranged from 0.14 ± 0.070 to $1.85\pm0.219~\mu g~^{-1}$ with the maximum content in the PA and the minimum content in the PC, which varied dramatically according to the ANOVA and Tukey tests (Additional file 1: Fig. S3D and Table S6).

To determine the difference between metals in plastic debris, the principal components analysis was performed to describe the principal components of primary metal among plastic types. Additional file 1: Table S7 summarizes the results of PCA for each factor. Two principal components with eigenvalues > 1 were taken into consideration. PC1 contributed to 64.63% variance with the highest eigenvalue 3.231, having the loading in As (R=2.549). The result of PC2 elucidated 31.94% of the total difference with the highest loading of Cu (R=1.463) (Fig. 5).

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Discussion

To our knowledge, this is the first report for systematically studying the size, shape, and color of plastic debris distributed on the Marina Beach in Chennai, India. In this study, we also characterized the pattern of absorbed metal elements in the plastic materials and quantified the concentration of As, Cr, Cu, and Pb on the plastics. As we know, the beach is a highly dynamic ecosystem, in which winds or waves always move sand and sediment [42]. The surrounding conditions of many beaches can lead to extensive changes in the composition of plastic debris. The higher number of beach plastic mainly comes from the presence of shops and hotels [43, 44]. In monsoonal months, the usage of the beach is less due to the rainfall. However, we found a high abundance of plastics in November 2017-December 2017, which is consistent with the report by Jayasiri et al. [24]. Previous studies indicated that plastics on the beach were much higher during the rainy season because of the surface runoff, which led the plastics scraped from inland through rivers and to settle on the coastal line near to estuaries [45–47]. Barnes stated that the abundance of plastics was very strongly related to the human population [48].

Petroleum-based plastics pollution in the Marina Beach

The collected plastics were categorized into two principal categories in shape and color. The first type of plastic debris in this study was fiber. The predominance of plastic fibers in coastal sediments was consistent with most of the previous findings [49-52]. Claessens et al. revealed the second most common type of plastic debris in Weizhou Island was the plastic fiber [49]. Plastic fragments had different types on the shores of the Adriatic, but many of the residues were jagged fragments of larger plastic materials. The primary types of shape (by number) were fragments around 60.6%, followed by 23.6% of film and 10.3% of fibers [34]. In our study, compared to other colors, the white plastic is the maximum abundant color. The percentage of white color plastic items has been detected in other studies, such as in Andaman Sea beach (Malaysia) 70% [53] and Atlantic Ocean beach (Southwest England) 59% [54]. 87.7% of the collected plastic was also found to be white [33, 55, 56]. Small debris between the low tide and vegetation lines of 5-25 mm were collected [57-59]. As far as we know, only a few studies reported mesoplastic debris on the field [13, 60].

Surface textures and validation of petroleum-based plastics

The result of the SEM analysis showed the pits and fractures in the plastics debris were generated by the chemical weathering. Besides, the chemical and mechanical

weathering of plastic contributed to the degradation of plastic particles. Our findings indicated that oxidative processes occurred in pits according to LDPE and PC. Small degradation patches in the PA showed the preferential site for continuous degradation, providing a favorable oxidation site. The adhering particles were the parts or pieces of other natural or synthetic substances grouped into larger components [61]. The cohesiveness of the surfaces results in adhesion to plastic particles. Analysis of petroleum-based plastics revealed that the adhering particles on PP and PC were prominent. The most common polymers found in the coastal area were the PET, PE, PS, and PP [38].

The FTIR spectroscopy provides a simple, effective, and non-destructive approach for the detection and identification of massive plastic polymers based on the well-known infrared absorbent database [62]. The PE and PP were found to dominate in the petroleumbased plastics in this study, because the PE and PP are the most produced plastic materials [30]. Commonly, the LDPE native bonds can be detected according to IR peaks at 1710 cm⁻¹, 1406 cm⁻¹, 1339 cm⁻¹, 1236 cm⁻¹, 1089 cm⁻¹, 1009 cm⁻¹, 866, and 717 cm⁻¹ [63]. The spectral bands of PET at 1710 cm⁻¹, 1406 cm⁻¹, 1339 cm⁻¹ 1236, 1089, and 1009 cm⁻¹, 866 cm⁻¹, and 717 cm⁻¹ were also previously described by Holland 2002 [64]. Our result was consistent with the report mentioned above. We also found the symmetric stretching of CH₂ at 2928 cm⁻¹ and the asymmetric stretching of CH₂ groups within the polymer chain. The C=O stretching was related to the spectrum at 1631.48 cm⁻¹ and a combination of the N-H band was at 1535 cm⁻¹. These peaks have been reported as the PA native bonds [65]. A strong intensity of carbonyl bonds C=O and an aromatic (C-H) at 2920 cm⁻¹ and 1768.4 cm⁻¹, indicating the PC native bonds [66].

The TGA–DSC analysis is a useful method for characterizing the physical properties of polymers, because it can be used for evaluating the melting point, crystallization, transition temperatures, and the changes of enthalpy and entropy [67]. In this study, the TGA–DSC curves were measured to detect endothermic reactions for all polymers. We hypothesized that the specific endothermic polymer peaks could be used to describe the petroleum-based plastic debris. The petroleum-based plastic debris (LDPE, PET, PA, and PC) showed the characteristic endothermic peaks [66, 68–70]. The results of ATR–FTIR and TGA–DSC confirmed a 100% rate of petroleum-based plastic debris.

The levels of trace metals in the petroleum-based plastics

Notably, the concentrations of metals in the plastics (Fig. 4) and metal concentration in fresh products

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(Additional file 1: Table S8) varied dramatically. The maximal levels occurred in plastic litter collected from sediments when compared to new products. Compared to the previously reported result, our collected plastic debris contained a higher amount of metals than sediments [20]. Plastics are the metal catcher and transporters to the sea [71], because it consists of organic polymer with high affinity to metal [17]. However, the absorption rate of metal to various plastic material is different due to the discrepancy in their physical nature and chemical properties [27, 72]. Metals can be accumulated in plastic debris from their surrounding ecosystem, which was confirmed by previous reports [20, 54, 73]. The plastics also can act as a vector for the transfer of trace metals into the food chain. Thus, birds, fish, mammals, and invertebrates may mobilize metals in their acidic digestive system, enzyme-rich pathways by mistakenly taking plastic material as food [12].

The content of As was high in the PA. The use of As in the manufacture formulations is now limited [74]. The high arsenic levels in the plastic debris may also be attributed to the age of manufactured objects resulting from the mesoplastic process or absorption [75]. In this study, the Cu concentrations in PET and PC were the lowest, but the sublethal dosage of Cu may also cause a significant reduction in the marine microalgae Tetraselmis chuii [8]. The level of Cu was low as reported in other studies [18]. The highest concentration of Cu in plastic fragments compared to virgin plastics was probably due to the higher surface area [76]. Cr has been detected in a similar level of magnitude in most plastics, except the PA. The plastic debris collected in the coastal areas had higher Cr content [17]. Cr is a ubiquitous ecosystem contaminant, which is toxic at ecologically relevant concentrations to the biota and humans. In this study, we found Pb in all-polymer debris, which can enter the food chain through biomagnification and bioaccumulation [77]. The significant difference of metal elements between plastic material was p < 0.05, according to Tukey's HSD post hoc test and Kruskal-Wallis test. In the Persian Gulf, higher concentrations of metals (Mn, Al, Fe, Cd, Cr, Ni, Pb, and Cu) in the sediments than those in plastics have been reported by Dobaradaran et al. [39].

Conclusions

Taken together, this study revealed different petroleum-based plastics in the Marina Beach of Chennai, India. The Marina Beach is commonly used for recreation, and the monsoon also contributes to carrying different plastic fragments from the waste piles into the lakes, streams, and rivers, eventually to the beach. The main shapes of petroleum-based plastic debris were fragments, spheres, and fibers. In this study, the color of petroleum-based

plastic debris was yellow, green, red, and blue. A total of 2275 petroleum-based plastic samples from the beach sediment were collected. The types of petroleum-based polymers, such as LDPE, PP, PET, PA, and PC were identified based on the ATR-FTIR and TGA-DSC analysis. For the mass determinations and type identification of plastic debris, the TGA-DSC provided an attractive option. LDPE, PP, PET, PA, and PC were quantified by the TGA-DSC. Furthermore, the ICP-MS analysis suggested that As, Cu, Cr, and Pb were released into the marine environment, posing a severe threat to the organism in the aquatic environment.

Supplementary information

Supplementary information accompanies this paper at https://doi. org/10.1186/s12302-020-00388-5.

Additional file1: Table S1. The content of different elements in the plastic material. **Table S2**. Analysis of the plastic number differences between the categories (S1, S2, S3) with a confidence interval of 95% using sites and Tukey (HSD). Table S3. Analysis of the Arsenic differences between the categories with a confidence interval of 95% using plastics and Tukey (HSD). Table S4. Analysis of the Chromium differences between the categories with a confidence interval of 95% using plastics and Tukey (HSD). Table S5. Analysis of the Copper differences between the categories with a confidence interval of 95% using plastics and Tukey (HSD). Table S6. Analysis of the Lead differences between the categories with a confidence interval of 95% using plastics and Tukey (HSD). Table S7. The principal component analysis of elements in the plastic debris. Table S8. Concentrations of metal carried by plastic debris in the environment. Fig. S1. Sample sites on the Marina Beach, Chennai, India. Fig. S2. The number and size of plastics in this study. Fig. S3. The metal concentration of different plastic debris collected from the Marina Beach, Chennai, India.

Abbreviations

As: Arsenic; Cr: Chromium; Cu: Copper; Pb: Lead; SEM: Scanning electron microscope; LDPE: Low-density polyethylene; PET: Polyethylene terephthalate; PP: Polypropylene; PA: Polyamide; PC: Polycarbonate; ICP-MS: Inductively coupled plasma mass spectrometry; ATR-FTIR: Fourier transform infrared spectroscopy—attenuated total reflectance; TGA-DSC: Thermogravimetry-differential scanning calorimetry.

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Authors' contributions

TYS and WGL were involved in the experiments and manuscript writing. AS, FR.PV, JAD, and JGM were responsible for data analysis. DSP was involved in conceptualization, funding acquisition, project administration, and manuscript revision. All authors read and approved the final manuscript.

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Availability of data and materials

The datasets obtained and analyzed during the current study are available from the corresponding author on reasonable request.

Ethics approval and consent to participate

Not applicable.

Consent for publication

Not applicable.

Competing interests

The authors declare that they have no competing interests.

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